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ways that defy traditional definitions for the stages of substance, reflecting the inherent complexity of science and the natural world, even when it appears as simple as determining whether a substance is a liquid or a solid. In this module we have focused on defining and explaining the basic properties of liquids, which for you Provides the foundation for thinking about the states of the case in all complexity. In other modules we discuss concrete and Steps to help you, unlike the different physical properties of these states. When it comes to different fluids, some mix well while others don't; Some pour quickly while others flow slowly. This module provides a foundation for considering the states of matter in all their complexity. It explains the basic properties of the fluid, and explores how intermolecular forces determine their behavior. The concepts of harmony, adhesion and viscosity are defined. The module also examines how the temperature and size and type of the molecule affect the properties of liquids. Key concepts share certain properties with liquid solids - both are considered condensed substances and are relatively uncompressed - and some with gases, such as their flow and the ability to shape their containers. Many properties of fluid, such as cohesion and adhesion, are influenced by intermolecular forces within the liquid itself. Viscosity is influenced by both the intermolecular forces and molecular shape of a compound. Most of the fluids that we encounter in everyday life are actually solutions, within the liquid solvent there is a mixture of solid, liquid or gas solute. HS-C6.2, HS-PS1. A3, HS-PS1. A4 Rachel Bernstein, PhD, Anthony Carpie, PhD Fluid Properties Visionlearning Vol. 3 (5), 2015. Top Top

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